

PRODUCT SAFETY  
DATA SHEET

CHEMICALS COMPANY

## A. GENERAL INFORMATION

DPM 950-25

TRADE NAME (COMMON NAME OR SYNONYM)

Oxalic Acid

☒ C.A.S. NO. ☐ ALLIED PRODUCT CODE #

144-62-7

CHEMICAL NAME

Oxalic Acid

FORMULA

 $(\text{COOH})_2 \cdot 2\text{H}_2\text{O}$ 

MOLECULAR WEIGHT

126.07

COMPANY/PLANT ADDRESS (No., STREET, CITY, STATE AND ZIP CODE)

Chemicals Company

P.O. Box 1139R

Morristown, N.J. 07960

CONTACT

Director - Technical Service

PHONE NUMBER

(315) 487-4990

ISSUED DATE

June 12, 1980

REVISED DATE

## B. FIRST AID MEASURES

Skin or Eyes: Promptly flush with plenty of water for at least 15 minutes.

EMERGENCY PHONE NUMBER  
(201) 455-2000

Ingestion: Drink large amounts of milk to dilute and neutralize the acid.  
Do not induce vomiting.

Get prompt medical attention for ingestion, inhalation, eye contact, irritation or burns.

## C. HAZARDS INFORMATION

## FIRE AND EXPLOSION

FLASH POINT

°C

AUTO IGNITION  
TEMPERATURE

°C

FLAMMABLE LIMITS IN AIR (% BY VOL.)

☐ OPEN CUP ☐ CLOSED CUP

NA - Not Applicable

LOWER NA

UPPER NA

UNUSUAL FIRE AND EXPLOSION HAZARDS

Partial decomposition occurs at 150° C. Decomposition products include carbon monoxide and formic acid, which are both toxic and flammable. Can react violently with strong oxidizers.

## HEALTH

INHALATION

Inhalation of dust or mists can cause irritation or burns to the upper respiratory system, including nose, mouth, and throat. Can also irritate lungs.

INGESTION

Can cause irritation and corrosive burns to mouth, throat, and stomach, etc. Can be fatal if swallowed. Oxalic acid is a systemic poison affecting the central nervous system and kidney function as well as other organs.

SKIN

Can cause irritation or corrosive burns. Dilute solutions can be irritating on prolonged exposure.

EYES

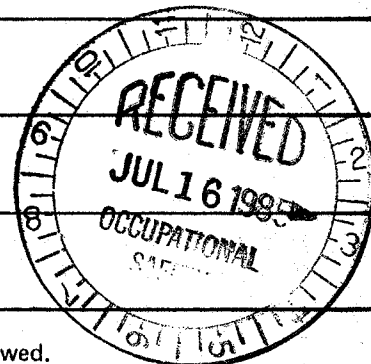
Can cause irritation, corneal burns, and conjunctivitis.

PERMISSIBLE CONCENTRATION: AIR Threshold Limit Value (TLV)  
(SEE SECTION J)

1 mg/m<sup>3</sup>

BIOLOGICAL

UNUSUAL CHRONIC TOXICITY



## D. PRECAUTIONS/PROCEDURES

### VENTILATION

Sufficient to reduce dust or mist concentrations below current permissible TLV levels.

### NORMAL HANDLING

Use appropriate precautions and protective equipment as outlined in Section E.

### STORAGE

Store in a cool, dry place away from food, food chemicals or strong oxidizers.

PRECAUTIONARY LABEL ☐ ATTACHED ☒ NOT ATTACHED

Label warning statement(s): "Danger - May be fatal if swallowed. Causes eye and skin irritation. Do not breathe dust. Keep away from food and food products."

### SPILL OR LEAK

Sweep and shovel up into container. Neutralize area with alkali solution. ( See Section I for disposal methods.)

### FIRE EXTINGUISHING AGENTS RECOMMENDED

NA

### SPECIAL FIRE FIGHTING PRECAUTIONS

Formic acid and carbon monoxide gases may be present. Self-contained breathing apparatus or air-supplied respirator may be required.

### FIRE EXTINGUISHING AGENTS TO AVOID

NA

### SPECIAL PRECAUTIONS/PROCEDURES

## E. PERSONAL PROTECTIVE EQUIPMENT

### RESPIRATORY PROTECTION

Where required, use a respirator approved by NIOSH for dusts or dusts and mists, as applicable.

### EYES AND FACE

As a minimum, wear hard hat and chemical safety goggles. Do not wear contact lenses. For further face protection, include full-face shield.

### HANDS, ARMS, AND BODY

As a minimum, wear long-sleeve shirt and trousers, boots and gloves for routine product use. Wear acid-resistant apron and impervious gloves when handling solutions.

### OTHER CLOTHING AND EQUIPMENT

## I. ENVIRONMENTAL

### DEGRADABILITY

5 day BOD: 0.12 lb/lb

### OCTANOL/WATER PARTITION COEFFICIENT

### WASTE DISPOSAL METHODS\*

Neutralize with alkali solution and flush to sewer with plenty of water if permitted by applicable disposal regulations. Neutralized waste may have to be disposed of by an approved contractor.

\*DISPOSER MUST COMPLY WITH FEDERAL, STATE AND LOCAL DISPOSAL OR DISCHARGE LAWS.

## J. REFERENCES

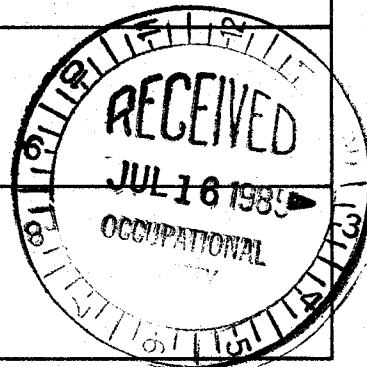
### PERMISSIBLE CONCENTRATION REFERENCES

OSHA standard, 29 CFR, Part 1910.1000 (July 1, 1977).

### REGULATORY STANDARDS

DOT Classification - Not Regulated (49 CFR).

### GENERAL



## K. ADDITIONAL INFORMATION

This product is not for food or drug use.



**LOS ANGELES CHEMICAL COMPANY**  
SOUTH GATE, CALIFORNIA 90280  
(213) 583-4761

THIS PRODUCT SAFETY DATA SHEET IS OFFERED SOLELY FOR YOUR INFORMATION, CONSIDERATION AND INVESTIGATION.

ALLIED CHEMICAL PROVIDES NO WARRANTIES, EITHER EXPRESS OR IMPLIED, AND ASSUMES NO RESPONSIBILITY FOR THE ACCURACY OR COMPLETENESS OF THE DATA CONTAINED HEREIN.

**F. PHYSICAL DATA**

<b>MATERIAL IS (AT NORMAL CONDITIONS):</b> <input type="checkbox"/> LIQUID <input checked="" type="checkbox"/> SOLID <input type="checkbox"/> GAS <input type="checkbox"/> _____		<b>APPEARANCE AND ODOR</b> Colorless crystals; Odorless.	
<b>BOILING POINT</b> °C	<b>SPECIFIC GRAVITY</b> (H <sub>2</sub> O = 1)	<b>VAPOR DENSITY</b> (AIR = 1)	
<b>MELTING POINT</b> 101.5 °C	1.653	NA	
<b>SOLUBILITY IN WATER</b> (% by weight)	<b>pH</b>	<b>VAPOR PRESSURE</b> (mm Hg at 20° C)	
13.7% solution at 25°C	1% solution*; pH = 1.3	NA	
<b>EVAPORATION RATE</b> (Butyl Acetate = 1)	<b>% VOLATILES BY VOLUME</b> (At 20° C)		
NA	NA	*anhydrous basis	

**G. REACTIVITY DATA**

<b>STABILITY</b>  <input type="checkbox"/> UNSTABLE <input checked="" type="checkbox"/> STABLE	<b>CONDITIONS TO AVOID</b>
<b>INCOMPATIBILITY (MATERIALS TO AVOID)</b> Reacts with alkalis, decomposes at high temperatures (above 150°C). Reacts with some silver compounds to form explosive silver oxalate. Can react violently with strong oxidizing materials.	
<b>HAZARDOUS DECOMPOSITION PRODUCTS</b>  formic acid, carbon monoxide.	
<b>HAZARDOUS POLYMERIZATION</b>  <input type="checkbox"/> MAY OCCUR <input checked="" type="checkbox"/> WILL NOT OCCUR	<b>CONDITIONS TO AVOID</b>

**H. HAZARDOUS INGREDIENTS (Mixtures Only)**

MATERIAL OR COMPONENT	%	HAZARD DATA (SEE SECT. J)
NA		